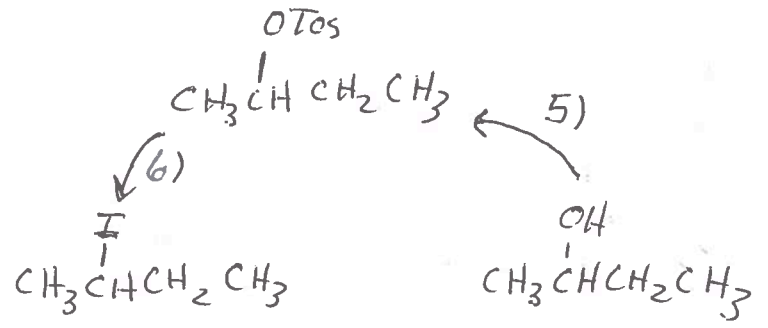


MAD ORG. CHEM "MIN." #5

NOT a useful sequence!

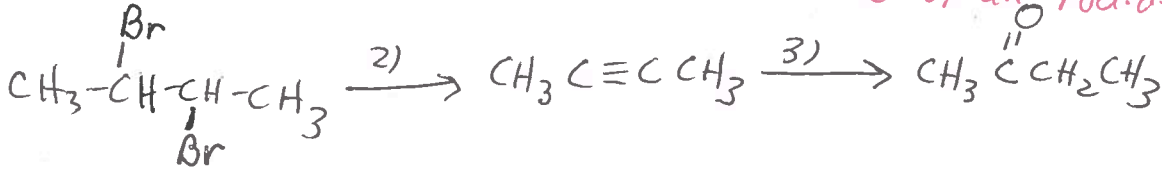
given for practice only!

1. CH3CH=CH2
  1. Br<sub>2</sub>/CCl<sub>4</sub>
  2. KOH/200 °C
  3. H<sub>2</sub>O/H<sub>2</sub>SO<sub>4</sub>/HgSO<sub>4</sub>
  4. NaBH<sub>4</sub>/EtOH

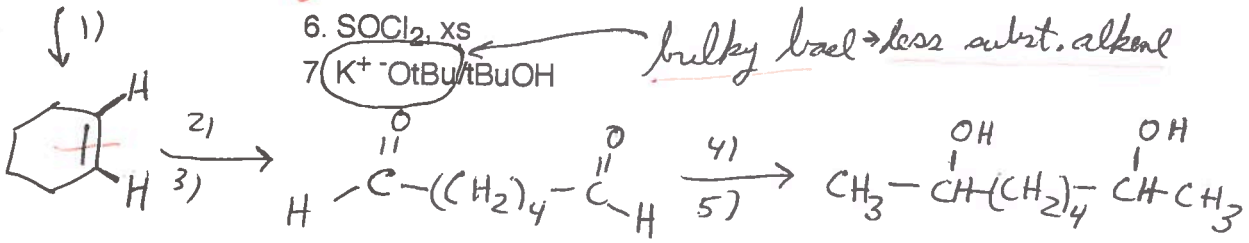
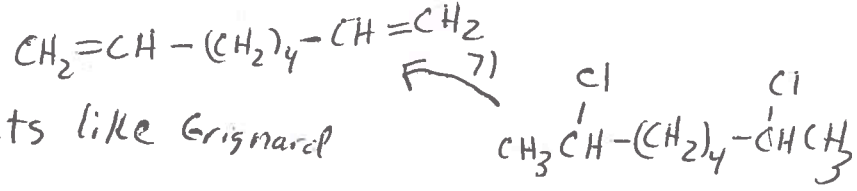


5. TsCl/pyridine
6. NaI/acetone

would we really go to this → No much trouble for an iodide? → No

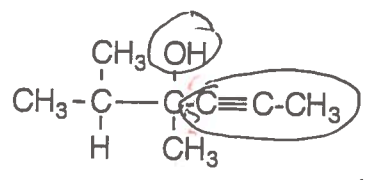


2. C1CCCCC1
  1. CH<sub>3</sub>O<sup>-</sup>Na<sup>+</sup>/CH<sub>3</sub>OH
  2. O<sub>3</sub>
  3. (CH<sub>3</sub>)<sub>2</sub>S
  4. CH<sub>3</sub>Li, 2 eq. ← reacts like Grignard
  5. H<sub>3</sub>O<sup>+</sup>

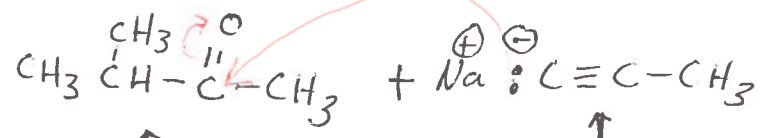


6. SOCl<sub>2</sub>, xs
  7. K<sup>+</sup> OtBu/tBuOH
- bulky base → less subst. alcohol

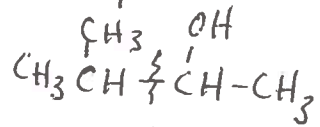
3. From any alkenes, alkynes or alcohols of three carbons or less, any inorganic reagents, and any oxidizing or reducing agents, synthesize the molecule below.



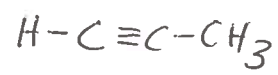
↑ then H<sub>3</sub>O<sup>+</sup>



↑ Jones ox



↑ NaNH<sub>2</sub>



↑ then H<sub>3</sub>O<sup>+</sup>

